

CHROMATOGRAPHIC PROFILING OF *Syzygium cumini* LEAVES USING HIGH-PERFORMANCE THIN LAYER CHROMATOGRAPHY AND GAS CHROMATOGRAPHY-MASS SPECTROMETRY TECHNIQUES

(Profil Kromatografi bagi daun *Syzygium cumini* menggunakan Teknik Kromatografi Lapisan Nipis Prestasi Tinggi dan Kromatografi Gas-Spektrometri Jisim)

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Abstract

Global use of traditional medicinal plants has increased logarithmically in the past few decades. Leaves of *Syzygium cumini* (*S. cumini*) contain important classes of pharmacologically active phytoconstituents like alkaloids, flavonoids, saponins, tannins, glycosides, phenols, fixed oils, monoterpenoids, steroids, and triterpenoids. Since community health is at risk due to improper knowledge and unenforced standardization parameters for herbal formulations, the focus of the study was to develop high-performance thin layer chromatography (HPTLC) fingerprinting and to identify phytoconstituents through gas chromatography and mass spectrometry (GC-MS). HPTLC fingerprinting of the methanolic extract of the leaves of *S. cumini* was performed using the mobile phase chloroform: methanol: ethyl acetate (6:4:6) after several pilot thin layer chromatography (TLC) analyses, which revealed 14 peaks at 254 nm and 10 peaks at 366 nm. Also, qualitative analysis of the same was performed using GC-MS, which revealed 9 phytoconstituents, some of them possessing known pharmacological activity. Such fingerprint analysis using sophisticated chromatographic and spectral techniques can contribute to the quality control in herbal industries; help in assuring purity, efficacy, and safety of the formulations by detection of adulteration; as well as broaden the horizon of phytochemical research, including the isolation of a marker therefrom.

Keywords: *Syzygium cumini*, adulteration, fingerprinting, gas chromatography mass spectrometry, high-performance thin layer chromatography, standardization

Abstrak

Penggunaan tumbuhan bagi perubatan tradisional secara global telah meningkat bagi tempoh beberapa dekad yang lepas. Daun *Syzygium cumini* (*S. cumini*) mengandungi jujukan fito aktif secara farmakologi merupakan kelas utama seperti alkaloid, flavonoid, saponin, tannin, glikosida, fenol, minyak, monoterpenoids, steroids, dan triterpenoids. Semenjak kesihatan komuniti berada pada keadaan berisiko disebabkan oleh maklumat yang kurang tepat dan tidak penguatkuasaan piawaian parameter bagi formulasi herbal, maka fokus kajian ini untuk membangunkan pencapjarian kromatografi lapisan nipis prestasi tinggi (HPTLC) dan mengenalpasti jujukan fito melalui kromatografi gas dan spektrometri jisim (GC-MS). Pencapjarian HPTLC terhadap

ekstrak metanol bagi daun *S. cumini* telah dijalankan menggunakan fasa bergerak klorofom: metanol: etil asetat (6:4:6) selepas beberapa analisis pilot kromatografi lapisan nipis (TLC), ia menjelaskan terdapat 14 puncak pada 254 nm dan 10 puncak pada 366 nm. Juga, analisis kualitatif telah dijalankan menggunakan GC-MS, dimana ia menjelaskan 9 jujukan fito, sebahagian dikenalpasti mempunyai aktiviti farmakologi. Analisis capjari menggunakan teknik kromatografi dan spektra boleh menyumbang kepada kawalan kualiti dalam industri herba; membantu dalam jaminan ketulenan, kejituan dan keselamatan dalam formulasi dengan cara mengenal pemalsuan; juga pengembangan bidang kajian fitokimia, termasuklah pemencilan penanda kimia.

Kata kunci: *Syzygium cumini*, pemalsuan, pencapjarian, kromatografi gas dan spektrometri jisim, kromatografi lapisan nipis prestasi tinggi, pemiawaian

Introduction

Syzygium cumini also known as *Eugenia cumini*, *Eugenia jambolana*, *Myrtus cumini*, *Syzygium jambolana*, *Syzygium jambolanum*, and *Eugenia caryophyllifolia*, is a tree belonging to the Myrtle family, Myrtaceae. It is found in India, Bangladesh, Nepal, Sri Lanka, and Myanmar and throughout south-east Asia and the Pacific Islands while it is also grown in Indonesia, Thailand, Australia, Malaysia, Kenya, Zimbabwe, Madagascar, Colombia, Zambia, Cuba, Mexico, Brazil, and some parts of the USA particularly Florida and Hawaii. It is commonly known as Indian blackberry, Java Plum, Jambolan while Surabhipatra (in Ayurveda); Jamun (in Hindi); Jambudo (in Gujarati); Brahaspati, Mahajambu, and Ksudrajambu (in Sanskrit)[1, 2, 3]. *S. cumini* leaves, shown in Figure 1, contain various classes of secondary metabolites: glycosides, alkaloids, triterpenoids, flavonoids, saponins, tannins, phenols, proteins, steroids, and fixed oils and fats. Major phytoconstituents of the leaves include sitosterols, crategolic (maslinic) acid, myricetin, quercetin, ellagic acid, and betulinic acid. Essential oils from leaves are rich in pinocarveol, cineole, α -terpeneol, ellagic acids, myrtenol, eucarvone, α -myrtenal, geranyl acetone, α -cadinol, kaempferol, pinocarvone, quercetin, isoquercetin [4, 5, 6].



Figure 1. Leaves of *Syzygium cumini*

Its leaves are found to be useful in diabetes, hepatitis, and dental disorders as well to be an anticancer, anti-diarrheal, cardioprotective, diuretic, gastroprotective, and hepatoprotective agent. The leaves possess anti-allergic, antipyretic, antibacterial, anti-inflammatory, antimicrobial, and antioxidant properties. The leaves are also used to treat stomachache, dermatopathy, leucorrhoea, constipation, fever, reduce DNA damage induced by radiation, and in inhibition of blood discharges in the feces [7-11].

S. cumini leaves are pharmacologically useful. However, the standardization of herbal formulations is always an issue. Additionally, government bodies have also levied certain regulations for the quality control and standardization of herbal products. However, chromatographic analysis of Indian medicinal plants using globally accepted sophisticated techniques is scarcely available.

This research may provide the HPTLC and GC-MS fingerprints of *S. cumini* leaves to herbal industries and society. It may facilitate standardization and quality control for the prevention of adulteration in *S. cumini* leaves formulations, thereby benefitting the end-user. It shall give an insight into the phytochemical composition of the leaves, which can be further worked upon.

Materials and Methods

Collection and sampling

Leaves of *S. cumini* were obtained from the Medicinal Garden of RK University in February (Figure 1). The sample was then authenticated by comparing it with standard literature. Its herbarium SOP/COG/91/2013 was prepared and submitted at the School of Pharmacy, RK University, and certified by the botanist of School of Science, RK University.

Extraction of *S. cumini* leaves

Leaves of *S. cumini* were dried in a hot air oven at 50 °C and powdered using a mixer grinder. Maceration of 50.0 g dry powder of the leaves with 100 mL methanol was carried out at room temperature for 24 hours. After filtration, on evaporating the excess solvent from the filtrate at room temperature, 7.2 g sticky, dry and solid extract was obtained.

Pilot TLC studies and HPTLC fingerprinting

Several pilot TLCs were developed for methanolic extract using various proportions of chloroform, methanol, and ethyl acetate. Highly non-polar petroleum ether extract and highly polar water extract were not used, as we had expected to obtain chemical constituents of medium polarity. After observing the results, many TLCs of methanolic extract was developed by adding ethyl acetate and formic acid for removal of tailing. The mobile phase Chloroform: Methanol: Ethyl acetate (6: 4: 6) showed the best resolution in TLC studies. Hence, it was further used for HPTLC fingerprinting.

Methanolic extract of *S. cumini* leaves was analyzed to obtain HPTLC fingerprinting using Chloroform: Methanol: Ethyl acetate (6: 4: 6) at the Department of Pharmaceutical Sciences, Saurashtra University, Rajkot, Gujarat, India. The HPTLC fingerprinting was obtained on HPTLC plates containing silica gel C₆₀ F₂₅₄ as stationary phase, manufactured by E.MERCK KgaA. CAMAG Linomat 5 “Linomat5_171010” S/N 171010 (2.01.02) was used for a sample application. Peak height and area were selected for evaluation, while the measurement was done using the principle of absorption. The sample was dissolved in Methanol. CAMAG TLC Scanner 3 “Scanner_171010” S/N 171010 (2.01.02) was used for scanning the plates in daylight as well as at 254 nm and 366 nm.

GC-MS analysis

The sample containing methanolic extract of *S. cumini* leaves was named Jambudo (local name of *S. cumini*). Qualitative GC-MS analysis of the sample was performed at the Department of Chemistry, School of Science, RK University, Rajkot, Gujarat, India. Agilent GC-7820A/MS-5977 with HP-5ms column containing (5%-phenyl)-methylpolysiloxane as stationary phase and helium as carrier gas was used for this purpose following the GC-MS protocol from authentic standard works of literature [12,13,14,15,16]. National Institute of Standards and Technology (NIST) library was used for the precise identification of phytoconstituents.

Results and Discussion

Pilot TLC studies and HPTLC fingerprinting

After several pilots, TLC analyses, an HPTLC fingerprinting of the methanolic extract of the leaves of *S. cumini* was performed using the mobile phase Chloroform: Methanol: Ethyl Acetate (6: 4: 6). The HPTLC plates were scanned at 254 nm (Figure 2), 366 nm (Figure 3), and daylight (Figure 4) that revealed 14 spots (Figure 5) at 254 nm and 10 spots (Figure 6) at 366 nm. The distinct R_f values of the spots at 254 nm and 366 nm are as provided in Table 1 and Table 2 respectively.

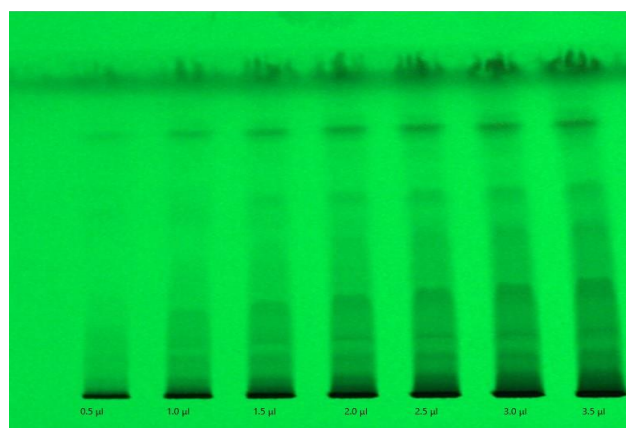


Figure 2. HPTLC plate at 254 nm

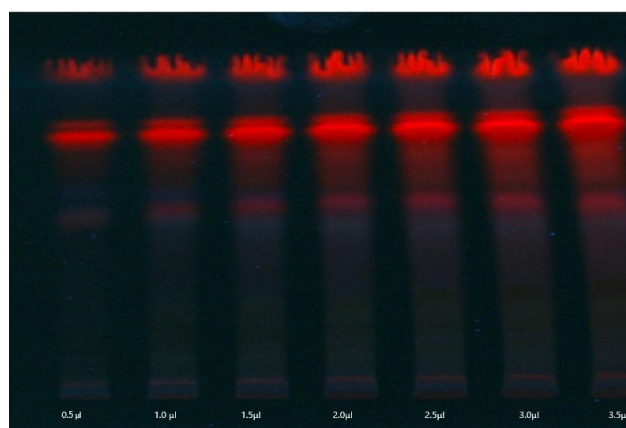


Figure 3. HPTLC plate at 366 nm

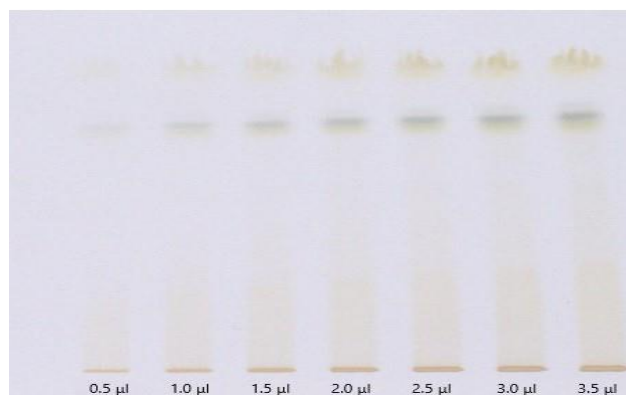


Figure 4. HPTLC plate at daylight

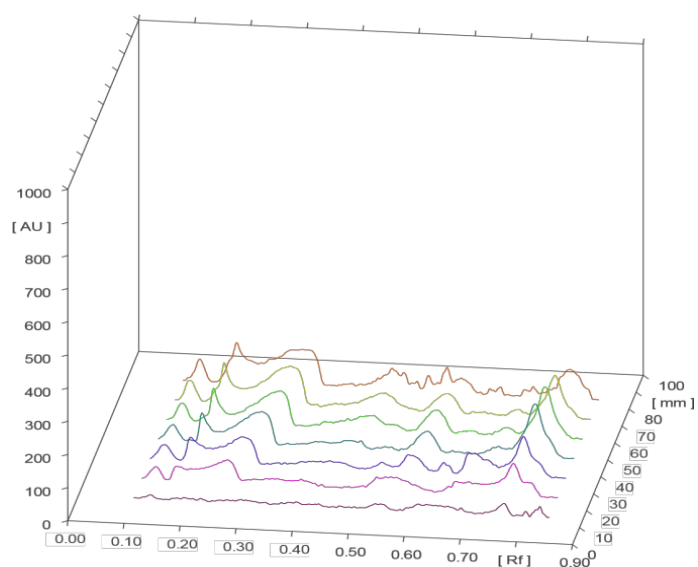


Figure 5. Densitometric chromatogram of methanolic extract of *S. cumini* leaves at 254 nm

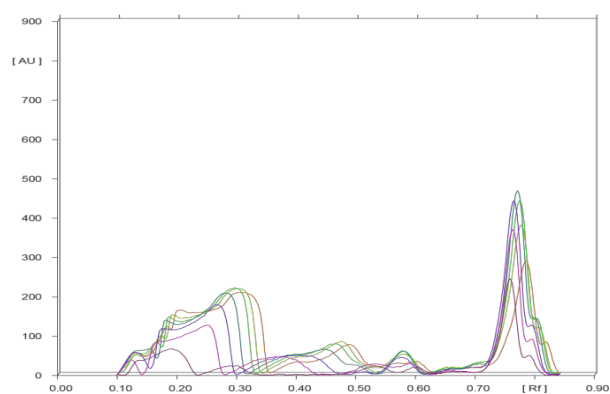


Figure 6. Densitometric chromatogram of methanolic extract of *S. cumini* leaves at 366 nm

Table 1. Maximum R_f and Area under curve at 254 nm

Peak	Max R_f	AUC ($\text{ng } \mu\text{L}^{-1}$)
1	0.13	1071.2
2	0.19	3841.1
3	0.31	6786.7
4	0.47	1727.1
5	0.49	548.9
6	0.52	367.7
7	0.54	535.1
8	0.57	1254.4
9	0.60	1081.0
10	0.65	167.0
11	0.67	218.9
12	0.71	188.0
13	0.72	314.9
14	0.79	3410.2

Table 2. Maximum R_f and Area under curve at 366 nm

Peak	Max R_f	AUC ($\text{ng } \mu\text{L}^{-1}$)
1	0.13	1304.5
2	0.18	2726.1
3	0.20	3751.8
4	0.30	15174.9
5	0.40	981.2
6	0.49	3995.0
7	0.60	735.7
8	0.67	298.7
9	0.78	9574.9
10	0.81	1127.4

GC - MS analysis

The GC-MS spectrum of methanolic extract of *S. cumini* leaves (Figure 7) revealed a confirmatory structure match of the nine chemical compounds when compared with the NIST database. The following compounds (as shown in Table 3) were identified from the analysis.

Compound 1: (1R)-2,6,6-trimethylbicyclo [3.1.1]hept-2-ene, Mol. formula - $\text{C}_{10}\text{H}_{16}$, Mol. wt. - 136.2340, RT - 2.31. GC-MS fragment: The peak at 2.31 minutes had a mass $[\text{M}^+]$ 136.2340. The daughter ion spectra of

compound 1 revealed the characteristic fragments m/z (as shown in Figure 8) 53, 55, 77.1, 78.1, 79.1, 80.1, 91.1, 92.1, 93.1, 105.1.

Compound 2: 2,5,5-trimethyl-cyclopentadiene; Mol. formula - C_8H_{12} ; Mol. wt. - 108.18, RT - 4.416. GC-MS fragment: The peak at 4.416 minutes had a mass $[\text{M}^+]$ 108.18. The daughter ion spectra of compound 2 revealed the characteristic fragments m/z (as shown in Figure 9) 51, 77, 78, 79, 91, 92, 93.1, 105, 107, 108.

Compound 3: α -terpineol; Mol. formula - $C_{10}H_{18}O$; Mol. wt. - 154.25; RT - 8.27. GC-MS fragment: The peak at 8.27 minutes had a mass $[M^+]$ 154.25. The daughter ion spectra of compound 3 revealed the

characteristic fragments m/z (as shown in Figure 10) 59.1, 77.1, 79, 81.1, 91.1, 92.1, 93.1, 94, 121.1, 136.1.

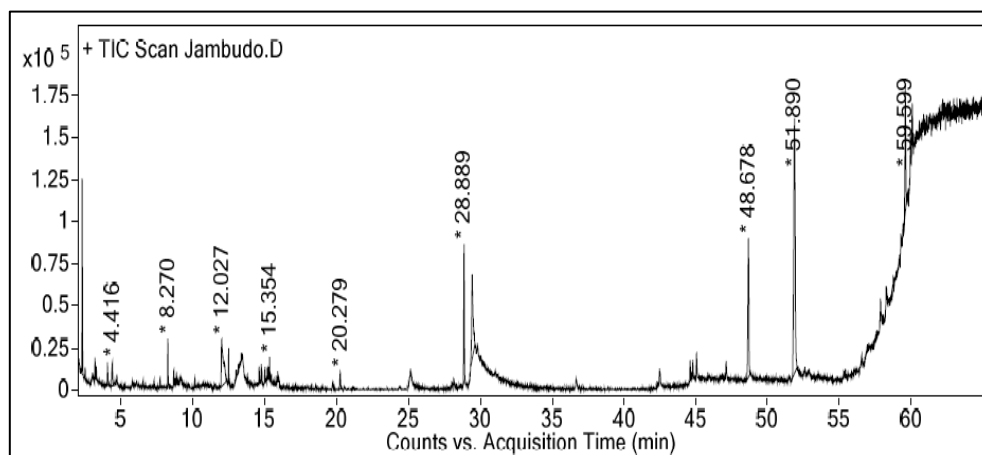


Figure 7. Gas chromatogram of methanolic extract of *S. cumini* leaves

Table 3. Phytoconstituents identified from methanolic extract of *S. cumini* leaves

Peak	RT	Name of Compound	Molecular Formula	Molecular Weight (gm/mol)	Peak Area %
1	2.31	(1R)-2,6,6-trimethylbicyclo[3.1.1]hept-2-ene	$C_{10}H_{16}$	136.2340	7.54
2	4.416	2,5,5-trimethyl-cyclopentadiene	C_8H_{12}	108.18	1.87
3	8.27	α -terpineol	$C_{10}H_{18}O$	154.25	1.95
4	12.027	1,2,3-benzenetriol	$C_6H_6O_3$	126.11	9.24
5	12.499	1,4-methanocycloocta[d]pyridazine,1,4,4a,5,6,9,10,10a-octa hydro-11,11-dimethyl-(1.alpha,4.alpha,4a.alpha,10a.alpha)	$C_{13}H_{20}N_2$	204.15	1.2
6	28.889	3,7,11,15-tetramethyl-2-hexadecen-1-ol	$C_{20}H_{40}O$	296.5310	7.28
7	29.451	9,12,15-octadecatrienoic acid, methyl ester (Z, Z, Z)	$C_{19}H_{32}O_2$	292.4562	10.45
8	48.678	Vitamin E	$C_{29}H_{50}O_2$	430.7061	13.23
9	51.89	γ -sitosterol	$C_{29}H_{50}O$	414.7067	34.01

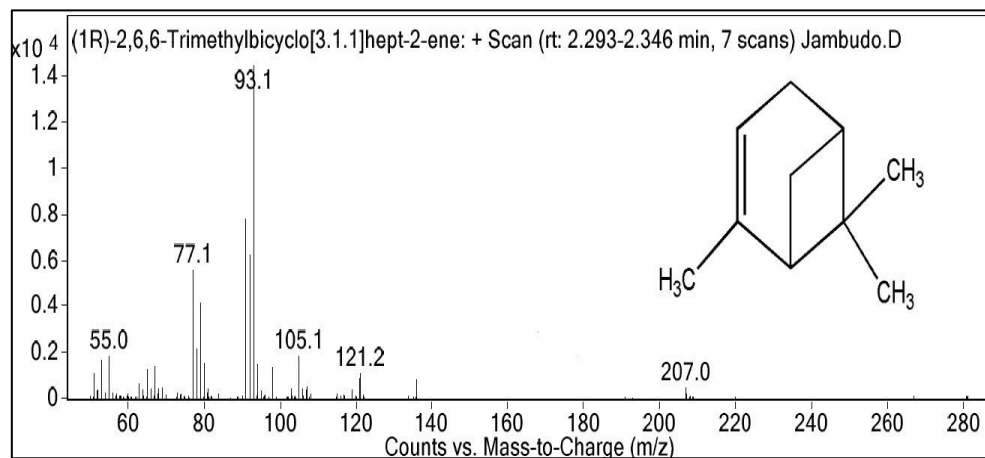


Figure 8. Mass spectrogram of Compound 1

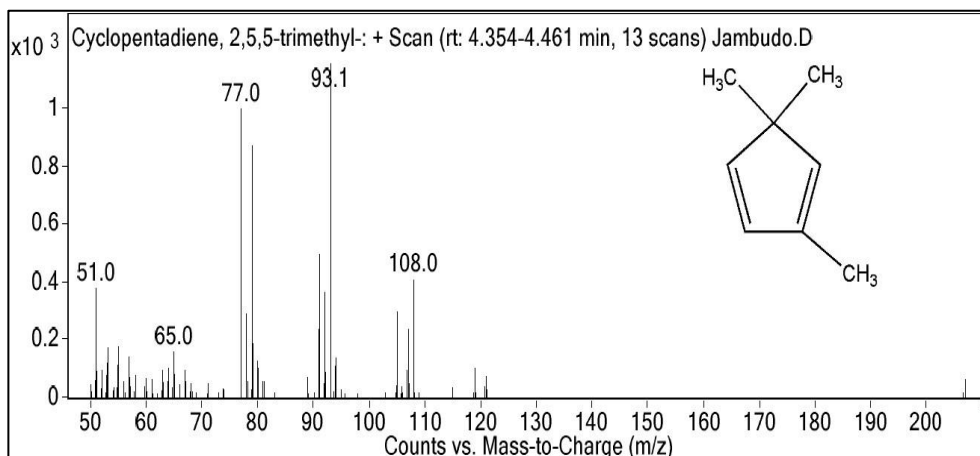


Figure 9. Mass spectrogram of Compound 2

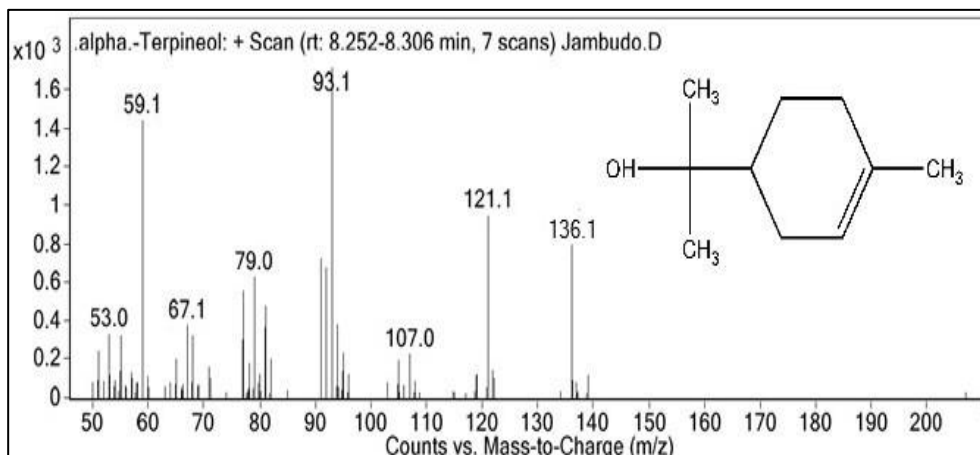


Figure 10. Mass spectrogram of Compound 3

Compound 4: 1,2,3-benzenetriol; Mol. formula - $C_6H_6O_3$; Mol. wt. - 126.11; RT - 12.027. GC-MS fragment: The peak at 12.027 minutes had a mass $[M^+]$ 126.11. The daughter ion spectra of compound 4 revealed the characteristic fragments m/z (as shown in Figure 11) 51, 52, 52.1, 52.9, 53.1, 79, 80, 97, 108, 126.

Compound 5: 1,4-methanocycloocta[d]pyridazine,1,4,4a,5,6,9,10,10a-octahydro-11,11dimethyl(1.alpha,4.alpha,4a.alpha,10a.alpha); Mol. formula - $C_{13}H_{20}N_2$; Mol. wt. - 204.15; RT - 12.499. GC-MS fragment: The peak at 12.499 minutes had a mass $[M^+]$ 204.15. The daughter ion spectra of compound 5 revealed the characteristic fragments m/z (as shown in Figure 12) 55.1, 67, 69.1, 77.1, 79.1, 91, 93.1, 105.1, 119.1, 133.

Compound 6: 3,7,11,15-tetramethyl-2-hexadecen-1-ol; Mol. formula - $C_{20}H_{40}O$; Mol. wt. - 296.5310; RT - 28.889. GC-MS fragment: The peak at 28.889 minutes had a mass $[M^+]$ 296.5310. The daughter ion spectra of compound 6 revealed the characteristic fragments m/z (as shown in Figure 13) 55.1, 57.1, 68.1, 69.1, 71.1, 79.1, 81.1, 82.1, 95.1, 123.1.

Compound 7: 9,12,15-octadecatrienoic acid, methyl ester (Z, Z, Z); Mol. formula - $C_{19}H_{32}O_2$; Mol. wt. - 292.4562; RT - 29.451. GC-MS fragment: The peak at 29.451 minutes had a mass $[M^+]$ 292.4562. The daughter ion spectra of compound 7 revealed the characteristic fragments m/z (as shown in Figure 14) 55.1, 60, 67.1, 77.1, 79.1, 80.1, 81.1, 91.1, 93.1, 95.1.

Compound 8: Vitamin E; Mol. formula - $C_{29}H_{50}O_2$; Mol. wt. - 430.7061; RT - 48.678. GC-MS fragment: The peak at 48.678 minutes had a mass $[M^+]$ 430.7061. The daughter ion spectra of compound 8 revealed the characteristic fragments m/z (as shown in Figure 15) 73, 121, 164, 164.1, 165.1, 166.1, 205.1, 430.4, 431.3, 431.4.

Compound 9: γ -sitosterol; Mol. formula - $C_{29}H_{50}O$; Mol. wt. - 414.7067; RT - 51.89. GC-MS fragment: The peak at 51.89 minutes had a mass $[M^+]$ 414.7067. The daughter ion spectra of compound 9 revealed the characteristic fragments m/z (as shown in Figure 16) 55.1, 57.1, 79, 81.1, 91.1, 95.1, 105.1, 145.1, 207, 396.3.

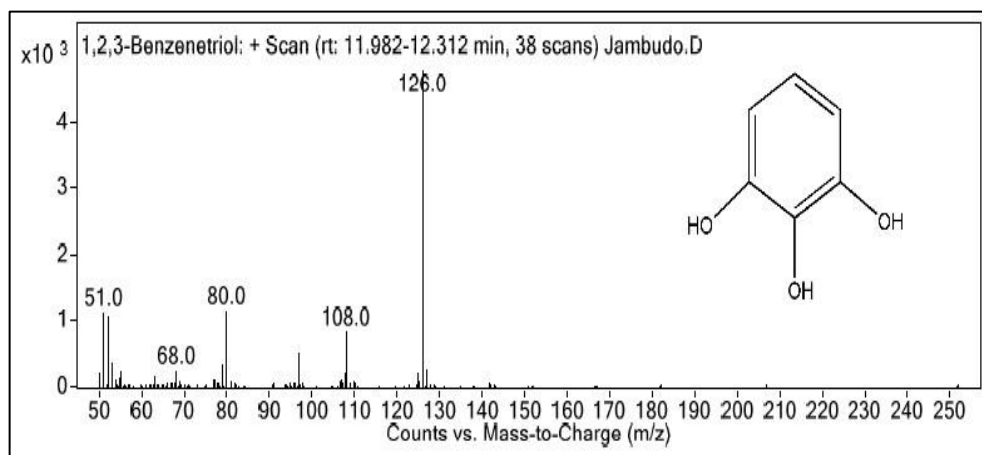


Figure 11. Mass spectrogram of Compound 4

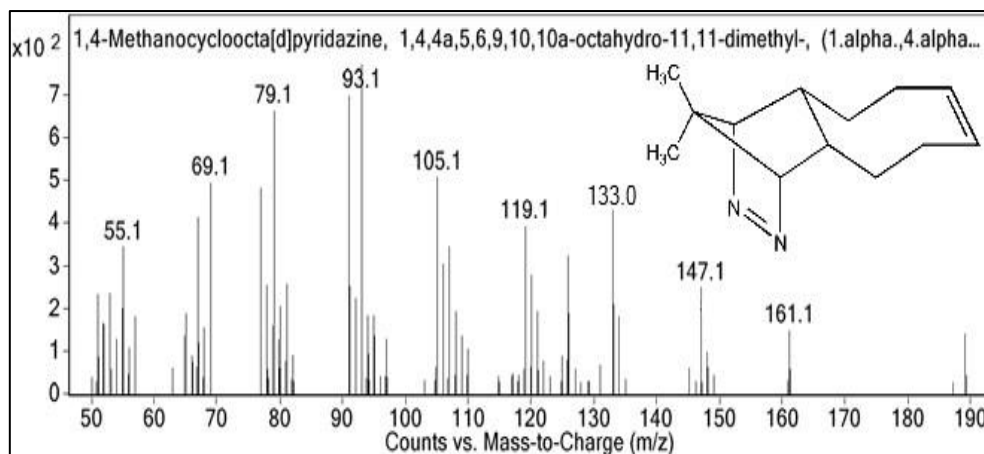


Figure 12. Mass spectrogram of Compound 5

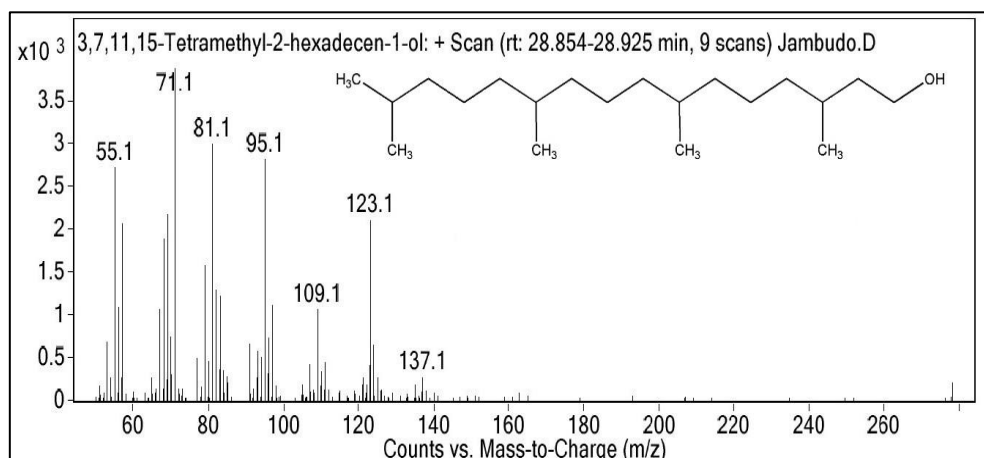


Figure 13. Mass spectrogram of Compound 6

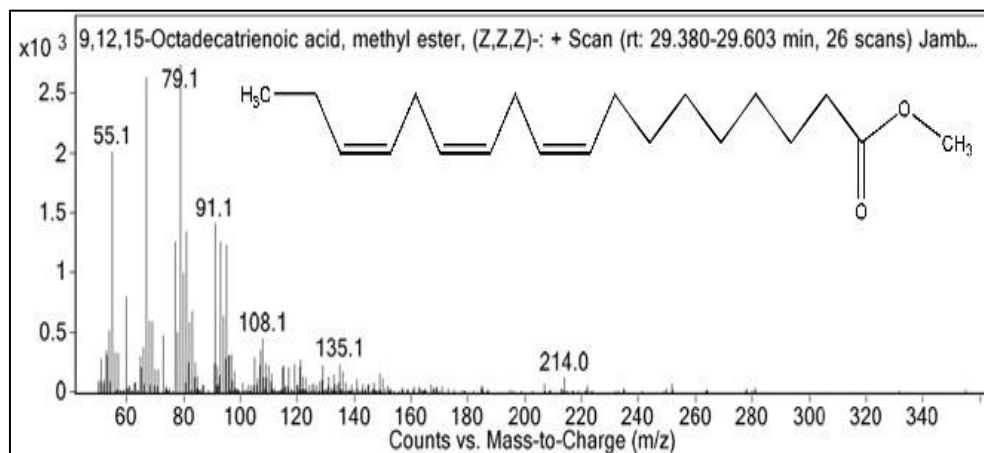


Figure 14. Mass spectrogram of Compound 7

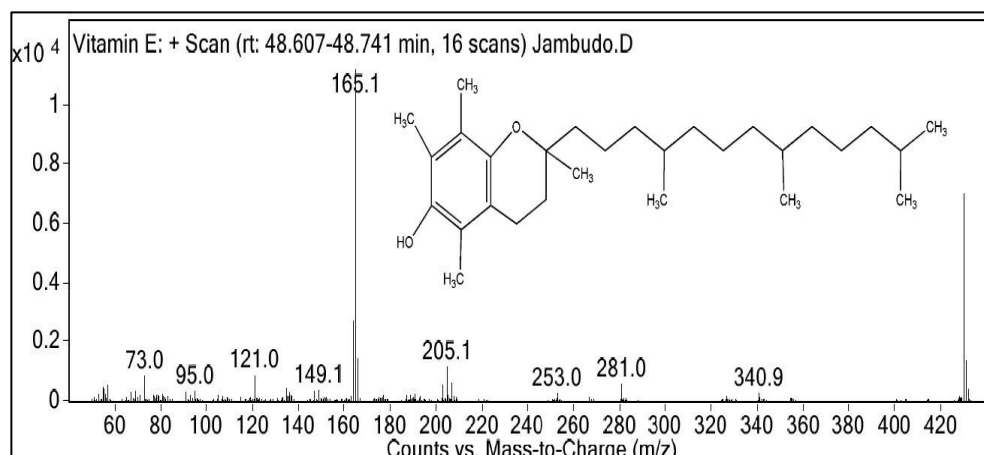


Figure 15. Mass spectrum of Compound 8

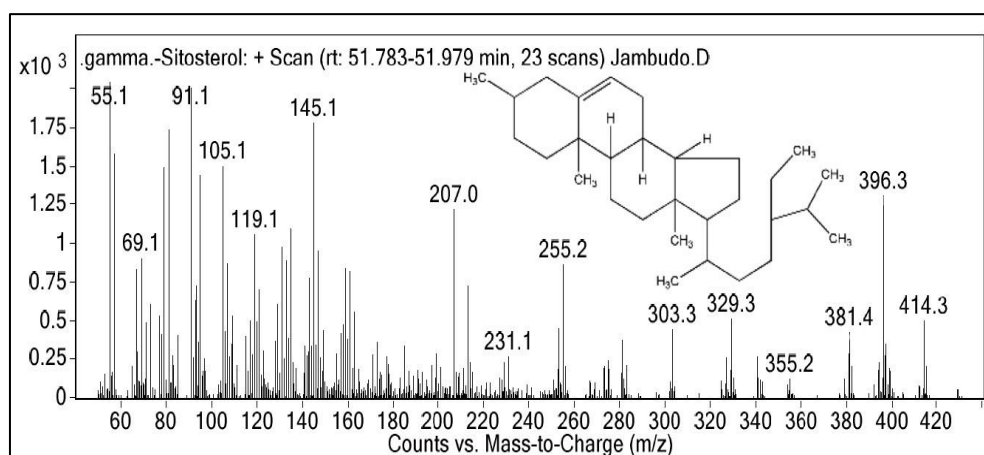


Figure 16. Mass spectrum of Compound 9

The following phytochemicals detected from the GC-MS analysis of a methanolic extract of *S. cumini* leaves (Table 4) were reported with significant biological activities:

It was found that many of the compounds detected in the GC-MS analysis possessed significant pharmacological activity. Few of them were antimicrobial and specifically antibacterial as well. The presence of Vitamin E explains a lot of health benefits

as given in the table. Some of them were found anticancer, anticonvulsant, antinociceptive, antioxidant as well as anti-inflammatory, while one of them was detected possessing antidiabetic activity.

It is noteworthy that Jaleel et al. (2019) have reported 10 different compounds through GC-MS analysis of *S. cumini* seeds extract in hexane [29]. In addition to that, Sharma et al. (2015) had identified and quantified 8 monoterpenoids by GC-MS in volatile oil of unripe

fruit pulp of *S. cumini*[30]. Chitnis et al. (2012) reported HPTLC analysis of available formulations containing *S. cumini* seeds using Toluene: Ethyl acetate: Formic acid (2: 7: 1) [31]. However, general HPTLC analysis by Kharat et al. revealed 12 peaks in ethanolic, 15 in methanolic, 14 in ethyl acetate, 11 in chloroform, and 3 in aqueous extract of *S. cumini* leaves using Toluene: Methanol: Diethylamine (30: 4: 4) solvent system using sonication method [32].

This exclusive method for HPTLC fingerprint and GC-MS analysis of *S. cumini* leaves may help to determine the quality of formulations as well as to detect adulteration by other materials in those formulations, thereby helping in standardization. It paves way for the isolation of the detected compounds. It also opens options for exploring other pharmacological uses of *S. cumini* leaves based on the reported biological properties of the detected compounds.

Table 4. Compounds detected from GC-MS of *S. cumini* leaves and their biological activities

Name of Compound	Reported Biological Activities
(1R)-2,6,6-trimethylbicyclo[3.1.1]hept-2-ene (commonly known as α -pinene)	✓ Anti-inflammatory[17] ✓ Antinociceptive[18] ✓ Inhibits hepatoma carcinoma cell proliferation[19]
α -terpineol	✓ Anticancer agent[20] ✓ Inhibits IL-1β, IL-6 & IL-10 production[21] ✓ Antimicrobial in dental problems[22]
3,7,11,15-tetramethyl-2-hexadecen-1-ol (commonly known as Phytol)	✓ Anticonvulsant[23] ✓ Antinociceptive and antioxidant[24]
9,12,15-octadecatrienoic acid, methyl ester, (Z, Z, Z) (commonly known as α -methyl linolenate)	✓ Antimicrobial and antibacterial[25] ✓ Downregulate melanogenesis[26]
Vitamin E (commonly known as α -tocopherol)	✓ Prevents loss of spermatogenesis and lipid peroxidation, antioxidant, inhibits smooth muscle proliferation, decreases protein kinase C activity, enhances prostacyclin release, upregulates cytosolic phospholipase A2 and cyclooxygenase activities, and facilitates natriuresis.[27]
γ -sitosterol	✓ Antidiabetic[28]

Conclusion

This preliminary fingerprint analysis of *S. cumini* leaves using sophisticated techniques may provide a starting point for quality control of its formulations; assuring purity, efficacy, and safety of the formulations; detecting adulteration; in standardization and ensuring isolation of a marker compound, thereby helping government regulatory agencies, herbal industries, and consumers. It may also broaden the horizon for further phytochemical and pharmacological

research, and the development of more sophisticated methods for standardization purposes.

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