



TWO ISOLATED CRYSTALLINE SOLIDS FROM THE STEM OF *Entada spiralis* Ridl. (AKAR SINTOK)

(Pemencilan Dua Jenis Pepejal Kristal dari Batang Pokok *Entada spiralis* (Akar Sintok))

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Abstract

Entada spiralis Ridl. (Leguminosae) is a woody climber which locally known as 'akar sintok'. The isolation and structure determination of two crystalline solids from active methanol fraction of *E. spiralis* stem bark had been examined. The structure identification of both solids was based on spectroscopic data (¹H and ¹³C NMR, HMQC, HMBC, DEPT135 and MS) along with comparison with literature data. The results revealed that these two crystalline solid sugars isolated from the most active fraction were known as penta-2-acetoxy-β-D-digitoxopyranosyl-(1→2)-fructofuranosyl-(6→4)-β-D-glucopyranosyl-(1→4)-glucopyranosyl-(1→2)-β-D-fructofuranosyl-(6→1)-β-D-glucopyranosyl-(4→1)-acetylglucosamine (**1**) and β-D-glucopyranosyl-(1→2)-β-D-glucopyranosyl-(1→3)-β-D-xylopyranosyl-(1→4)-α,L-rhamnopyranosyl-(1→3)-β-D-glucopyranosyl-(1→3)-β-D-glucopyranoside (**2**). Thus, this finding can be utilized as a scientific baseline information for further skin disease studies, since both constituents were isolated from moderate antidermatophytic active fraction.

Keywords: *Entada spiralis*, crystalline saponin, glycosides, isolation

Abstrak

Pemencilan dan penentuan struktur dua komponen bersifat kristal dari batang pokok *E. spiralis* yang juga dikenali sebagai akar sintok telah dijalankan. Eksperimen penentuan struktur berdasarkan data – data spektroskopi menggunakan ¹H, ¹³C NMR, HMQC, HMBC, DEPT135 dan spektrometer jisim dan juga perbandingan dengan kajian terdahulu. Hasil analisa spektroskopi yang terperinci menunjukkan kedua – dua kristal tersebut dikenali sebagai penta-2-asetoksi-β-D-digitoxopiranosil-(1→2)-fruktofuranosil-(6→4)-β-D-glukopiranosil-(1→4)-glukopiranosil-(1→2)-β-D-fruktofuranosil-(6→1)-β-D-glukopiranosil-(4→1)-asetilglukosamina (**1**) and β-D-glukopiranosil-(1→2)-β-D-glukopiranosil-(1→3)-β-D-xilopiranosil-(1→4)-α,L-rhamnopiranosil-(1→3)-β-D-glukopiranosil-(1→3)-β-D-glukopiranosid (**2**). Penemuan ini boleh dijadikan maklumat asas untuk kajian penyakit kulit kerana kedua-dua komponen ini dipencilkan dari fraksi metanol yang bersifat antiderma.

Kata kunci: *Entada spiralis*, kristal saponin, glikosid, pemencilan

Introduction

Medicinal herbs have been consumed traditionally by people throughout the world to cure various diseases long enough before the emerging of modern medicine. Due to their higher content of such valuable chemical compounds, they could be used to prevent microbial invasion. Secondary metabolites produced by plants contain a tremendous source of bioactive compounds such as flavonoid, alkaloid, tannin, saponin and terpenoid. Knowing their

importance, scientific investigation has increased rapidly to overcome the development of resistant pattern of certain microorganisms such as skin disease caused microbes [1].

The structure identification of bioactive constituents from the various types of species of Leguminosae family had been reported [2-6]. The stem of *E. spiralis* is believed to exhibit antimicrobial properties since it is capable to treat hair problem, itching and body cleansing. Previous phytochemical investigation of other *Entada* species encountered oleanolic acid, echinocystic acid, entagenic acid and acacic acid glycosides Okada et al. [7, 8]. *E. rheedii* Spreng was found to possess saponins [7-10], thiomides [11] and phenylacetic derivatives [12-13]. Investigation also revealed entagenic acid isolated from seed kernels of *E. rheedii* which exhibited moderate cytotoxic potency and antioxidative properties [14]. Phenolic and flavonoid were isolated from the root of *E. Africa* [15], octadecanoic acid from the stem bark of *E. abyssinica* [16] and Entadoiside from kernel nut of *E. phaseoloides* [17].

In this present study, we embark upon isolation and structure identification of two novel crystalline solids from active fractions of *E. spiralis* stem bark.

Materials and Methods

General Experimental Procedures

The general experimental procedures were adapted from Aiza et al. [18]. The ^1H Nuclear Magnetic Resonance (NMR) and ^{13}C NMR spectra were recorded in CDCl_3 , Fourier Transform NMR (FT-NMR) analysis was done using a Cryoprobe on a Bruker Avance 111 600 MHz spectrometer. Tetramethylsilane (TMS) was used as an internal standard in NMR spectrometer analysis. Electrospray ionization mass spectrometer (ESIMS) was performed using a Bruker micrOTOF-Q 86 mass spectrometer operating in positive-ion mode. Vacuum liquid chromatography (VLC) was conducted using silica gel 60 with a 230 – 400 mesh particle size (Merck). Thin layer chromatography (TLC) was performed on Kieselgel 60 F254 (Merck) aluminium support plates developed using chloroform: methanol system. TLC plates were visualized at UV_{366} and UV_{254} .

Fractionation and Isolation

The extraction and fractionation procedures were conducted as previously reported with few modifications [19, 20]. The extraction process involved consecutive soaking with petroleum ether, chloroform and methanol, fractionation process was conducted using vacuum liquid chromatography with the use of several binary solvents to obtain fractions.

Antidermatophytic Evaluation

The antidermatophytic activity of fractions was determined as previously reported by Aiza et al [21]. Generally, all extracts were used against three types of skin diseases dermatophytes namely as *Trychophyton tonsurans*, *Trychophyton mentagrophytes* and *Microsporum gypseum* through disc diffusion method susceptibility testing. The effectiveness of the fractions was evaluated by means of inhibition zone (IZ) measurement around the disk.

Results and Discussion

Compound (**1**) was obtained as white crystalline solid after subjecting methanol extract to vacuum liquid chromatography with an elution system of CHCl_3 -MeOH (6:4 v/v) to obtain fraction 1 which was previously found to possess moderate antidermatophytic activity [20]. The melting point of white crystalline solid was 187°C and two ultra violet (UV) absorption peaks appeared at 228 nm and 280 nm when analyzed with UV VIS-1800 Series (Shimadzu, Japan). The results indicated the absorption of light in the region of 200 – 300 nm, corresponded to the presence of atom with non-bonding orbital such as oxygen and nitrogen. Two peaks observed in the UV spectra were due to the presence of the chromophores, molecules that absorb light with bathochromic effect.

The Electron Spray Ionization Mass Spectrum (ESIMS) of compound (**1**) showed an $[\text{M}+\text{H}]^+$ ion peak at m/z 1905 in accordance with the molecular formula of $\text{C}_{78}\text{H}_{123}\text{NO}_{52}$. The fragment ion peaks were observed on the ESIMS at m/z 1049 $[\text{M}+\text{H}-171-172-171-171-171]$, 365 $[\text{M}+\text{H}-171-172-171-171-171-342-342]$ corresponding, to the consecutive loss of five pentosyls, two hexoxyls (Glc and Fru) and again two hexoxyls (Glc and Fru) respectively. Table 1 and Figure 1 showed the fragmentation of sugar moieties in the whole structure.

Table 1. Fragmentation of ions in compound **(1)** by ESIMS

Mass	Fragmentation of ions
1734	$[M+H] - 171]^+$
1562	$[M+H] - 171-172]^+$
1391	$[M+H] - 171-172-171]^+$
1220	$[M+H] - 171-172-171-171]^+$
1049	$[M+H] - 171-172-171-171-171]^+$
707	$[M+H] - 171-172-171-171-171-342]^+$
365	$[M+H] - 171-172-171-171-171-342-342]^+$

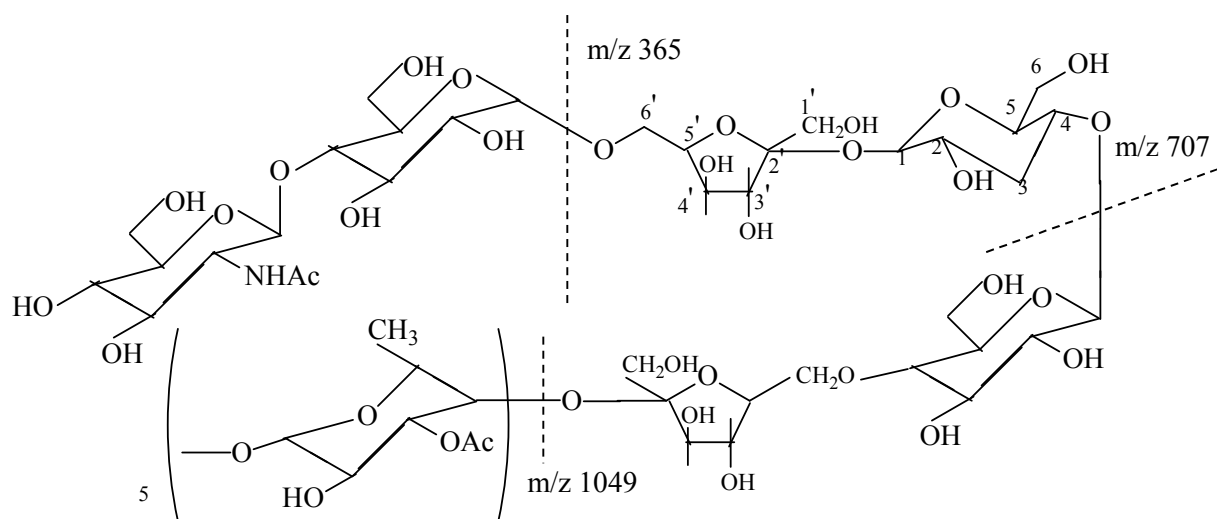


Figure 1. The mass fragmentation of compound **(1)** from *E. spiralis* stem bark

The ^1H NMR spectrum of compound **(1)** displayed three sugar anomeric protons at chemical shifts, ppm δ 4.41, 5.18 and 4.80, which indicated the presence of sugar moieties GlcNAc, Glc, acetyl digitoxose, respectively. These values were quite close to those reported in literature. The δ_{H} for GlcNAc was reported [22] as 4.40 ppm, δ_{H} 5.44 and δ_{H} 5.0 for acetyl digitoxose [23]. The proton NMR spectrum furnished nine hydroxyl protons at chemical shifts δ_{H} 3.14, 3.20, 3.47, 3.49, 3.55, 3.58, 3.68, 3.78 and 3.90, giving direct coupling with nine carbon atoms at δ 70.30, 72.09, 73.27, 60.95, 62.59, 83.01, 73.33, 74.74 and 77.47 ppm, respectively based on HMQC spectrum. The correlation between δ_{H} 5.18 and chemical shifts of carbon at δ_{C} 92.21 corresponded to the direct correlation of anomeric proton and anomeric carbon of Glc.

According to ^{13}C NMR spectrum, the chemical shifts at δ_{C} 104.49 and δ_{C} 92.21 suggested the presence of anomeric carbon for fructose (Fru) and glucose (Glc) respectively, and were quite close to those reported for Fru (δ_{C} 104.0) and Glc (δ_{C} 93.9) [10, 21]. The high intensity of overlapped peaks indicated more than one sugar present and this was same for absorption peaks of carbon hydroxyl. Thus, it may explain why compound **(1)** showed high molecular weight as displayed in the mass spectrum.

Long range correlations observed in the HMBC spectrum between signals at δ_H 5.18 (Glc-I) and δ_C 104.49 (C2'-Fru) showed three bond correlations between Glc and Fru. The four bond correlation was observed between anomeric proton of GlcNAc δ 4.4 and δ_C 73.27 (C3-Glc-II) suggested the location GlcNAc in AC1 structure. The anomeric proton of Glc-II (δ_H 5.18) was linked with Fru by four bond correlations at C5' (δ_C 83.01) of Fru. The spectrum also depicted three bond correlations between anomeric proton δ_H 4.80 of acetyldigitoxose moiety and δ_C 104.49 (C2'-Fru) which showed the linkage to Fru. The two bond correlations can be seen between signal at δ_H 3.90 (C5-glc-I) and δ_C 62.69 (C6-glc-I), δ_H 3.47 (C3-glc-I) and δ_C 72.09 (C2-glc-I) and δ_C 70.30 (C4-glc-I), δ_H 3.14 (C4-glc-I) and δ_C 73.33 (C3-glc-I), δ_H 3.68 (C3'-Fru) and δ_C 74.74 (C4'-Fru) and δ_C 83.01 (C5'-Fru). The three bond correlations were depicted between signal at δ_H 3.68 (C3'-Fru) and δ_C 83.01 (C5-Fru) and δ_C 74.74 (C4'-Fru), δ_H 3.49 (C1'-Fru) and δ_C 73.33 (C3'-Fru), δ_H 5.18 (C1-Glc) and δ_C 70.30 (C4-Glu), δ_H 3.58 (C5'-Fru) and δ_C 104.49 (C2'-Fru). The structure of sugar moieties and the HMBC correlation are shown in Figure 2 and 3 respectively. The proton and carbon NMR data of compound (1) in DMSO were summarized in Table 2.

Table 2. 1H NMR (600 MHz) and ^{13}C NMR (150 MHz) data of AC1 in DMSO

Position	DEPT	^{13}C (δ) (ppm)	1H δ (ppm), m	HMBC
Glu				
1	CH	92.21	5.19, m	C2', C4,C5
2	CH	72.09	3.20, m	
3	CH	73.27	3.47, m	C2,C4
4	CH	70.30	3.14, m	C4',C3
5	CH	77.47	3.90, t	C6
6	CH ₂	62.59	3.55, m	
Fru				
1'	CH ₂	60.95	3.49, m	C3'
2'	C	104.49	-	
3'	CH	73.33	3.68, m	C4',C5'
4'	CH	74.74	3.78, m	C2'
5'	CH	83.01	3.58, m	C2'
6'	CH ₂	62.51	3.43, m	
GlcNAc				
1	CH		4.41, m	C3
Acetyl digitoxose				
	CH		4.80, m	C2'

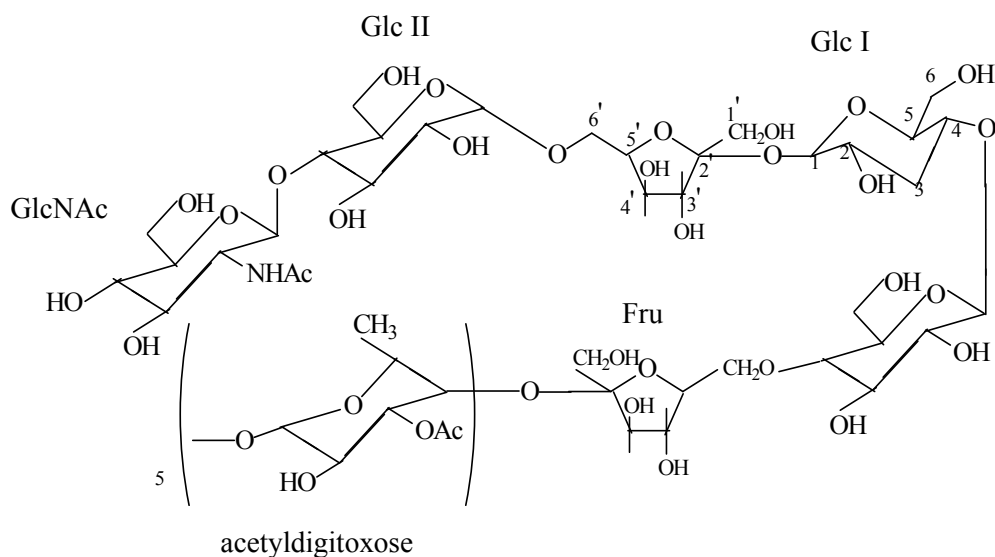


Figure 2. The structure of compound **(1)** from *E. spiralis* stem bark

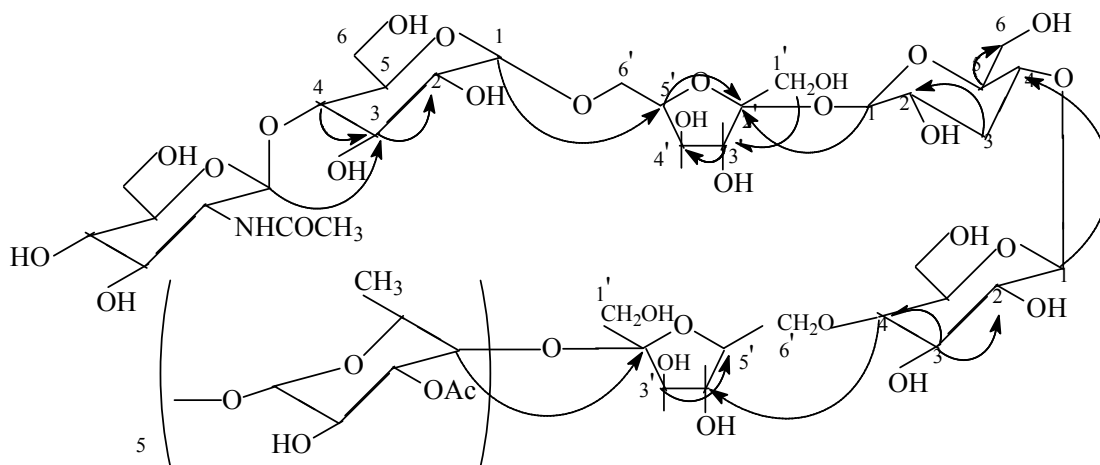


Figure 3. The key HMBC correlation of compound **(1)** from *E. spiralis* stem bark

After gone through an extensive spectroscopic analysis and comparison with literature, compound **(1)** was identified as penta-2-acetoxy- β -D-digitoxopyranosyl-(1 $\rightarrow$ 2)-fructofuranosyl-(6 $\rightarrow$ 4)- β -D-glucopyranosyl-(1 $\rightarrow$ 4)-glucopyranosyl-(1 $\rightarrow$ 2)- β -Dfructofuranosyl-(6 $\rightarrow$ 1)- β -D-glucopyranosyl-(4 $\rightarrow$ 1)-acetylglucosamine.

Compound **(2)** was isolated as white crystalline solid after subjecting to vacuum liquid chromatography with an elution system of CHCl_3 -MeOH (7:3 v/v), whereby fraction obtained was found to exhibit moderate antidermatophytic activity. The UV absorption analysis has displayed two absorption peaks at 256 and 280 nm which indicated chromophore of C=O group. The ESIMS displayed $[\text{M}+\text{H}]^+$ ion peak at m/z 922.0135, supported a molecular formula of $\text{C}_{35}\text{H}_{59}\text{O}_{28}$. The fragment ion peaks were observed on the ESIMS at m/z 600.5184 $[\text{M}+\text{H}-325]^+$ corresponding to the consecutive loss of two hexosyls. The m/z 325.1126 $[\text{M}+\text{H}-325-278]^+$ referred to the loss of

one pentosyl and again one hexosyl. Further fragmentation ion peak is observed at m/z 163.0607 $[M+H-325-278-162]^+$, which showed the loss of another one hexosyl. The mass fragmentation is illustrated in Figure 4.

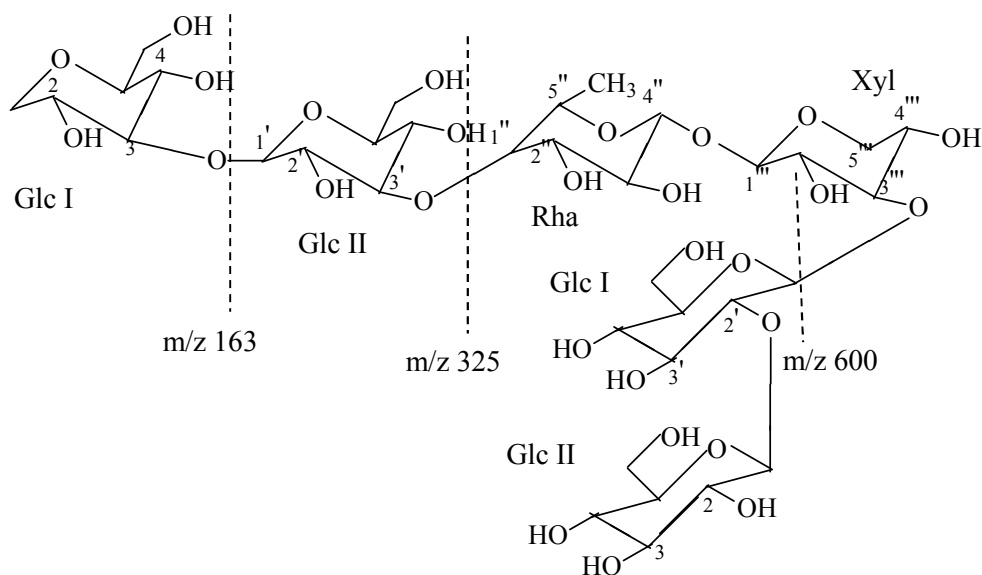


Figure 4. The mass fragmentations of compound **(2)** from *E. spiralis* stem bark

The sequences of the sugar chain were confirmed by HMBC experiments. The cross-peak observed in the HMBC spectrum between anomeric proton at signal δ_H 4.4 (Glc-I) and δ_C 83.02 (C2'), δ_H 5.05 (Rha-1) and δ_C 73.28 (C3'), δ_H 4.78 (Xyl) and δ_C 73.33 (C4''), δ_H 5.20 (Glc-II) and δ_C 73.28 (C3), showed the location of sugar moieties.

Other correlations with two bond correlations can be observed between proton signal δ_H 3.58 (C2-glc I) and δ_C 73.28 (C3), and δ_H 3.58 (C2'-Glc II) with δ_C 73.28 (C3'-Glc II). Three bond correlations can be seen between proton signal δ_H 3.43 (C5''-Rha) and δ_C 70.30 (C2''), 3.55 (C5'''-xyl) and δ_C 74.75 (C4''') and between δ_H 5.20 (C1'-Glc II) and δ_C 77.48 (C3'''-Xyl). The HMBC correlation is shown in Figure 5 and Table 3 showed the NMR data of compound **(2)**.

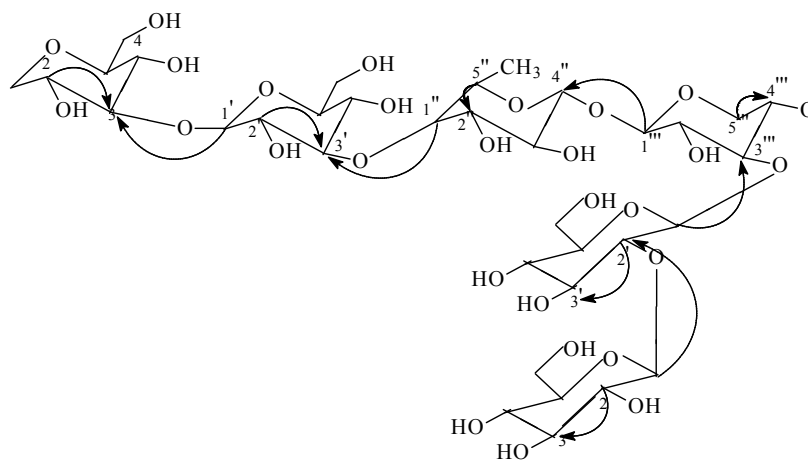


Figure 5. The key HMBC correlation of compound **(2)** from *E. spiralis* stem bark

After went through an extensive spectroscopic analysis and comparison with literature, compound **(2)** was elucidated as β -D-glucopyranosyl(1 $\rightarrow$ 2)- β -D-glucopyranosyl-(1 $\rightarrow$ 3)- β -D-xylopyranosyl(1 $\rightarrow$ 4)- α -L-rhamnopyranosyl-(1 $\rightarrow$ 3)- β -D-glucopyranosyl(1 $\rightarrow$ 3)- β -D-glucopyranoside as shown in Figure 6.

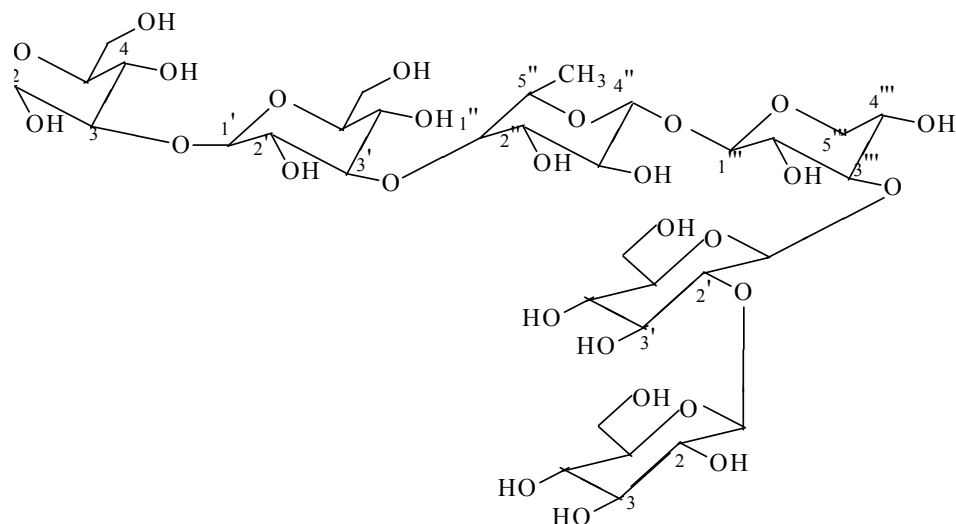


Figure 6. The structure of compound **(2)** from *E. spiralis* stem bark

Table 3. ^1H NMR (600 MHz) and ^{13}C NMR (150 MHz) data of compound **(2)** in DMSO

Position	DEPT	^{13}C (δ) (ppm)	^1H δ (ppm), m	HMBC
Glc I				
1	CH	92.21	4.4, m	C2'
2	CH	83.02	3.58, m	C3'
3	CH	73.28	3.47, m	
4	CH	70.30	3.14, m	
5	CH	77.47	3.90, t	
6	CH ₂	62.59	3.55, m	
Glc II				
1'	CH	92.21	5.20, m	C3, C3'''
2'	CH	83.02	3.58, m	C3'
3'	CH	73.28	3.47, m	
4'	CH	70.30	3.14, m	
5'	CH	77.47	3.90, m	
6'	CH ₂	62.51	3.43, m	

Table 3 (cont'd). ¹H NMR (600 MHz) and ¹³C NMR (150 MHz) data of compound (2) in DMSO

Position	DEPT	¹³ C (δ) (ppm)	¹ H δ (ppm), m	HMBC
Rha				
1''	CH	-	5.05, m	C3'
2''	CH	70.30	3.14, m	
3''	CH	72.09	3.20, m	
4''	CH	73.33	3.68, m	
5''	CH	62.51	3.43, m	C2''
6''	CH ₃	-	2.5, t	
Xyl				
1'''	CH	104.49	4.78, m	C4''
2'''	CH	74.75	3.78, m	
3'''	CH	77.48	3.90, m	
4'''	CH	70.30	3.14, m	
5'''	CH ₂	60.95	3.49, m	C4'''

Conclusion

Two types of crystalline solids from *E.spiralis* stem bark have been successfully isolated and characterized and they were determined as penta-2-acetoxy-β-D-digitoxopyranosyl-(1→2)-fructofuranosyl-(6→4)-β-D-glucopyranosyl-(1→4)-glucopyranosyl-(1→2)-β-D-fructofuranosyl-(6→1)-β-D-glucopyranosyl-(4→1) acetylglucosamine (1) and β,D-glucopyranosyl(1→2)-β,D-glucopyranosyl-(1→3)-β,D-xylopyranosyl(1→4)-α,L-rhamnopyranosyl-(1→3)-β,D-glucopyranosyl(1→3)-β,D-glucopyranoside (2). Both compounds have been isolated from methanol extract which is antidermatophytic active constituents.

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